

Cracking the Puzzle of CO₂ Formation on Interstellar Ices

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The formation of an ice layer on top of interstellar dust grains is a central topic in astrochemistry. The chemical composition of interstellar ices varies over time [1], and the dominance of polar or apolar fractions in them is important for several surface properties, like adsorption, diffusion or reaction. Among the most abundant components of interstellar ice we find H₂O, CO, CH₃OH or CO₂, to give some examples. These molecules can be formed *in situ* on interstellar ices, as it is the case of H₂O or CH₃OH, or can land on the ices after formation in the gas, like CO. Experiments show that CO₂ mainly forms *via* surface reactions (see, for example [2]), e.g. the CO + OH → CO₂ + H reaction. In this talk I will review our recent calculations on such reaction [3]. I will cover the reactivity on a CO and H₂O ice, showing that the reaction does not proceed directly in neither of them. In reality, the reaction stops at the very stable HOCO radical that may be later transformed into CO₂ via a H abstraction reaction. The H₂O ice is found to be a better substrate for the formation of HOCO, which can be behind the higher abundance of the polar component of CO₂ found in astronomical observations, in comparison with the apolar one [4].

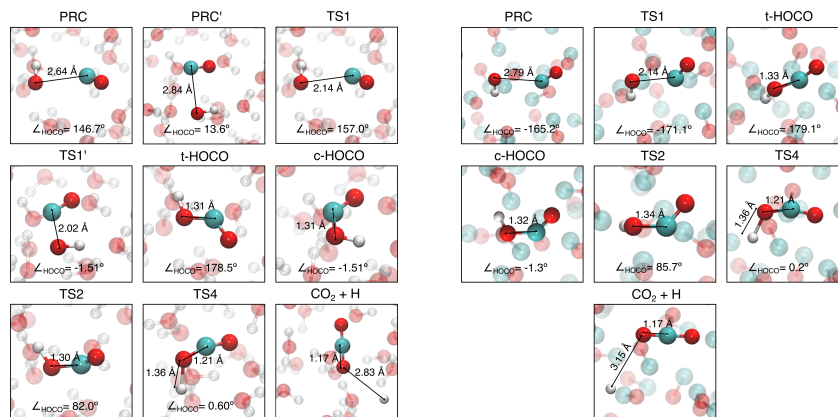


Figure 1: Stationary points in the potential energy surface for the CO + OH → CO₂ + H reaction on water (left) and CO ice (right).

References

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