

# IR spectroscopy and energetic processing of methyl isocyanate ice

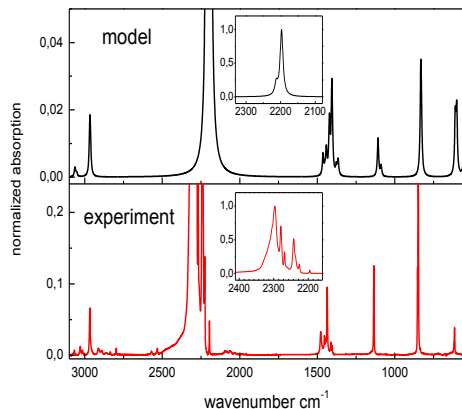
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Methyl isocyanate was recently detected in the interstellar medium [1-2] (the reported detection on comet 67P/CG [3] could not be confirmed[4]). Although it was usually neglected in astrochemical networks, this new evidence fosters new studies on this species. We present here our work on the IR spectra and on the energetic processing of ices of this molecule.

We have recorded in our lab the IR spectra of amorphous  $\text{CH}_3\text{NCO}$  and  $\text{CD}_3\text{NCO}$  obtained by vapor deposition at 20 K. The most intense band corresponds to the NCO asymmetric stretching ( $\nu_a\text{-NCO}$ ). It has a characteristic quadruplet structure, usually attributed to the interaction with the  $\text{CH}_3$  torsion.  $\text{CD}_3\text{NCO}$  was used to help the spectral assignment. We have also measured the band strengths for the absorptions of  $\text{CH}_3\text{NCO}$  in ice at 20 K.

Since no X-ray structure for crystalline  $\text{CH}_3\text{NCO}$  had been reported, we put forward a tentative theoretical structure, derived taking as a starting point the crystal of isocyanic acid. We have predicted a spectrum for the proposed structure, and compared it with our experimental observation [5] (see Fig. 1).



**Fig. 1.** Comparison of theoretical spectrum of the crystal with experiment.

We have studied the stability of  $\text{CH}_3\text{NCO}$  under heating, UV irradiation with a  $\text{D}_2$  lamp (UV photons  $\approx 10.3\text{-}6.9$  eV), and cosmic ray bombardment, simulated with high energy electrons (5 keV) [6]. The samples were either pure  $\text{CH}_3\text{NCO}$  or  $\text{CH}_3\text{NCO}$  diluted in  $\text{H}_2\text{O}$ . Heating of the ices leads to variations in the  $\nu_a\text{-NCO}$  band profile but not to an appreciable depletion of  $\text{CH}_3\text{NCO}$  until  $\text{H}_2\text{O}$  sublimation (beyond  $T=150$  K) and thus rules out hydrolysis in the ice. UV irradiation and electron bombardment at 20 K lead to the destruction of  $\text{CH}_3\text{NCO}$  and to the formation of  $\text{CO}$ ,  $\text{CO}_2$  and  $\text{OCN}^-$  ions in the processed ices.

These results indicate that the lifetime of hypothetical  $\text{CH}_3\text{NCO}$  present in the ice mantles of dust grains in a typical dense cloud might be longer than the cloud's lifetime.

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## References

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